

## Syllabus

### Course description

<b>Course title</b>	Electrochemical energy storage and conversion
<b>Course code</b>	45534
<b>Scientific sector</b>	ING-IND/23 "Applied Physical Chemistry"
<b>Degree</b>	Master Energy Engineering
<b>Semester</b>	2
<b>Year</b>	2
<b>Academic year</b>	2022/2023
<b>Credits</b>	6
<b>Modular</b>	no
<b>Total lecturing hours</b>	60
<b>Total lab hours</b>	
<b>Total exercise hours</b>	
<b>Attendance</b>	Exercise lessons are mandatory
<b>Prerequisites</b>	General chemistry- Physics: thermodynamics and electromagnetism
<b>Course page</b>	<a href="https://www.unibz.it/courses/45534">Course Offering / Free University of Bozen-Bolzano (unibz.it)</a>
<b>Specific educational objectives</b>	<p>Knowledge of principles of equilibrium and non-equilibrium electrochemistry and of the main electrochemical energy conversion and storage methods. The course introduces the fundamental notions of electrochemistry and surface thermodynamics necessary for a basic understanding of the physical chemical bases of electrochemical phenomena; how electrochemical and photo-electrochemical devices work. The main industrial electrochemical technologies dedicated to the energy production and storage are presented and analyzed through lectures, laboratory experiments and, if possible, visits to industrial plants.</p>
<b>Lecturer</b>	Dr. Ataollahi Narges
<b>Scientific sector of the lecturer</b>	ING-IND/23
<b>Teaching language</b>	English
<b>Office hours</b>	By appointment
<b>Teaching assistant (if any)</b>	-
<b>Office hours</b>	-
<b>List of topics covered</b>	<p>General definitions of electrochemical systems; equilibrium and non-equilibrium phenomena in electrochemistry; electrochemical kinetics; energy devices in electrochemistry: batteries, fuel cells, capacitors and</p>

	supercapacitors; photovoltaic cells and photo-electrochemical cells.
<b>Teaching format</b>	Frontal lectures and exercises
<b>Learning outcomes</b>	<p><b>Knowledge and understanding:</b> profound and detailed scientific knowledge and understanding of the principles of equilibrium and non-equilibrium electrochemistry</p> <p><b>Applying Knowledge and understanding:</b> profound and detailed scientific knowledge of the main electrochemical energy conversion and storage methods</p> <p><b>Making judgments:</b> skills and problem solving capacity to analyze problems of electrochemical energy conversion and storage</p> <p><b>Communication skills:</b> ability to structure and prepare scientific and technical documentation describing project activities</p> <p><b>Learning skills:</b> ability to independently keep up to date with developments in the most important areas of electrochemical energy conversion and storage</p>
<b>Assessment</b>	Evaluation of the written reports on the lab experiments and oral examinations. The examination consists of either an oral test, or the discussion of a report written by the student on a topic which may be either freely chosen or chosen in agreement with the course lecturer. An alternative method for the exam is building an electrochemical device with given specifications (e.g. A zinc-carbon battery which provides a given amount of power for a given time) and discussion of the procedures used and the problems found.
<b>Assessment language</b>	English
<b>Evaluation criteria and criteria for awarding marks</b>	Showing a sufficient knowledge of the topic and the ability to answer related questions
<b>Required readings</b>	<p>One of the following books: Bianchi e Mussini-Elettrochimica - ed. Masson Bianchi – Processi elettrochimici - ed. Masson D'Archer e Hill (Eds)</p> <p>Fundamentals of electrochemistry Bagotsky</p> <p>Electrochemistry for material science - Plieth</p> <p>Other files dedicated to specific topics will be indicated or given during the course.</p>
<b>Supplementary readings</b>	Other files dedicated to specific topics will be indicated or given during the course.